# LIQUIZYME POTASSIUM (Colorimetric Method)

### Code Product Name Pack Size LS033A Liquizyme Potassium 25 T LS033B Liquizyme Potassium 50 T LS033D Liquizyme Potassium 1 X 120 ml

# Intended Use

Diagnostic reagent for quantitative in vitro determination of Potassium in human serum.

# **Clinical Significance**

Potassium is estimated by Turbidimetic Method Potassium ions in the specimen react with sodium tetra phenyl boron to produce an insoluble potassium tetra phenyl boron resulting in a turbid suspension. The extent of turbidity is proportional to the potassium concentration and is measured photometrically at 620 nm (610-620).

# Principle

The amount of Potassium is determined by using sodium tetra phenyl boron in a specifically prepared mixture to produce a colloidal suspension. The turbidity of which is proportional to potassium concentration in the range of 2 - 7 mEq/L. Tetra phenyl Boron + K+ ' White Turbidity

# **Reagent Composition**

Reagent 1: Potassium Reagent Sodium tetra phenyl borate >60 mmol/L

Reagent 2 : Potassium Standard : 5 mEq/L Ready to use

# Materials Required But Not Provided

- Clean & Dry container.
- Laboratory Glass Pippetes or Micropioettes & Tips
- Colorimeter or Bio-Chemistry Analyzer.

# **Reagent Preparation**

Reagent is liquid, ready to use.

# **Stability And Storage**

The unopened reagents are stable till the expiry date stated on the bottle and kit label when stored at 25-30°C.

## Specimen Collection And Handling

Use unheamolytic serum. It is recommended to follow NCCLS procedures (or similar standardized conditions).

# Stability:

Do not use lipaemic / turbid / icteric samples. Discard contaminated specimens.

# Calibration

Calibration with the Potassium standard provided in the kit is recommended.

# Quality Control

It's recommended to run normal and abnormal control sera to validate reagent performance.

# **Expected Values**

: 3.5-5.5 mEg/L Serum

It is recommended that each laboratory verify this range or derives reference interval for the population it serves.

# Performance Data

Data contained within this section is representative of performance on Beacon system. Data obtained in your laboratory may differ from these values.

Limit of quantification	:	0.3 mEq/L
Linearity	:	8.5 mEq/L
Measuring range	:	0.3 - 8.5mEq/

# Precision

Intra-assay precision	Mean	SD	CV
Within run (n=20)	(mEq/L)	(mEq/L)	(%)
Sample 1	4.48	0.03	0.69
Sample 2	6.25	0.01	0.20
Inter-assay precision	Mean	SD	CV
Run to run (n=20)	(mEq/L)	(mEq/L)	(%)
Sample 1	6.23	0.016	0.25

# Comparison

A comparison between Beacon Potassium (y) and a commercially available test (x) using 20 samples gave following results :

1.008 x - 0.103 = ٧ 0.999 = r

# Warning And Precautions

- 1. For in vitro diagnostic use.
- 2. Specimens should be considered infections and handled appropriately.
- 3. Avoid ingestion. Do not pipette by mouth.
- 4. The reagent contains sodium hydroxide that is corrosive. In case of contact skin, flush with water. For eyes, seek medical attention.

# Waste Management

Please refer to local legal requirements.

Assay Procedure		
Wavelength	:	630 nm
Cuvette	:	1 cm

Pipette	into	clean	dry	test	tubes	labeled	as	Blank	(B),
Standa	rd (S)	and Te	st (T	):					

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Addition Sequence	Reagent Blank	Standard	Sample
Reagent 1	1000 µl	1000 µl	1000 µl
Standard	-	20 µl	-
Sample	-	-	20 µl
Distilled Water	20 µl	-	-

Mix and incubate for 5 minutes at room temperature. Measure the absorbance of the standard absorbance and sample absorbance against the reagent blank, at 630 nm.

# Calculation

Potassium (mEq/L) =  $\frac{Abs. T}{Abs. S} \times 5$ 

# **Assay Parameters For Photometers**

Mode	End point
Wavelength 1 (nm)	630
Sample Volume (µl)	20
Reagent Volume (µl)	1000
Incubation time (min.)	5
Incubation temp. (°C)	Room Temperature
Normal Low (mEq/L)	3.5
Normal High (mEq/L)	5.5
Linearity Low (mEq/L)	0.3
Linearity High (mEq/L)	8.5
Standard Concentration	5 mEq/L
Blank with	Reagent
Unit	mEq/L

# References

- Cornall, A. G., Bardawill, C. J., David, M. M.: J. Biol. Chem. 177, 751, 1949.
- 2.Doumas, B.T., Bayse, D.D.akol.: Clin. Chem. 27, 1642, 1981.
- 3. Chromy, V., Fischer, J.: Clin. Chem. 23, 754, 1977.
- 4. Chromy, V., Fischer, J., Voznieek, J.: Z. Med. Labor.-Diagn. 21, 333, 1980.
- 5.Tietz Textbook of Clinical Chemistry and Molecular diagnostics. Burtis, C.A.,
- 6.Ashwood, E.R., Bruns, D.E.; 5th edition, WB Saunders.

# Symbols Used On Labels

